Synthesis and Biological Activity of 4a,4"-Disubstituted Avermectins

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Allylic oxidation of avermectin B_1 and $C_{4''}$ -substituted avermectin B_1 analogues at C_{4a} followed by derivatization of the C_{4a} oxygen provides compounds that have an improved safety profile, as judged by mouse LD_{50} data, and are substantially more potent than avermectin B_1 against flies.

Keywords: Avermectin; anthelmintic; pesticide

INTRODUCTION

The avermectins are naturally occurring macrocyclic lactones with important anthelmintic and pesticidal activity (Fisher and Mrozik, 1984; Davies and Green, 1986; Pulliam et al., 1989; Lankas and Gordon, 1989; Davies and Green, 1991a,b). Avermeetin B_1 (abameetin) (Dybas, 1989) and 22,23-dihydroavermectin B_1 (ivermectin) (Chabala et al., 1980) are widely used anthelmintic agents in animal (Benz et al., 1989) and human health (Greene et al., 1989). The structural novelty of avermectins and their economic importance have attracted the attention of a number of synthetic organic chemistry groups concerned with chemical modification and total synthesis (Blizzard et al., 1990). In this vein, several publications and patents have appeared describing modification of avermectin B_1 at C_{4a} (Mrozik, 1984a,b) and $C_{4''}$ (Mrozik et al., 1989; Mrozik, 1984a,b; Linn and Mrozik, 1991) in an attempt to improve these important antiparasitic drugs.



Avermectin B_1 is isolated as a two-component mixture. The major component $(B_{1a}, \geq 80\%)$ contains a secbutyl group at C_{25} ; the minor component $(B_{1b}, \leq 20\%)$ contains an isopropyl group at C_{25} . These components are separable by HPLC; however, since they have essentially the same biological activities, they are not separated in practice. All compounds in this paper are mixtures of B_{1a} and B_{1b} components, represented as B_1 and drawn as B_{1a} for clarity.

We report in this paper our observations on the synthesis and biological activity of C_{4a} -substituted aver-

mectin analogues, with emphasis on their activity against flies. We have found that oxidative substitution at C_{4a} increases potency against flies and increases the LD_{50} in mice of these avermectin analogues.

EXPERIMENTAL PROCEDURES

General. General procedures and instrumentation employed in this work have been reported previously (Jones et al., 1992) except for the following additions. NMR assignments were made with the aid of APT data and literature data (Diez-Martin et al., 1990; Neszmelyi et al., 1989). Standard abbreviations are used for multiplicities with the addition of om for obscured multiplet. Low-resolution FAB mass spectra were acquired on Finnigan MAT TSQ 70B and VG Masslab VG20-250 quadrupole mass spectrometers using as matrix 5:1 dithiothreitol/dithioerythritol doped with lithium acetate or undoped. High-resolution LSIMS data were obtained on a JEOL JMS-HX110 double-focusing (EB) mass spectrometer at 10 kV using Ultramark 1600F as the internal standard and 3-nitrobenzyl alcohol as matrix.

Stomoxys Assay. The antifly activity was estimated in vitro by evaluating the response of adult Stomoxys calcitrans allowed to feed on 2.3 cm diameter Whatman Grade 3 filter paper disks containing bovine blood and test avermectins. Each compound was presented at five or more dosage levels in 0.1 mL of the experimental vehicle (aqueous vehicle containing 0.02% Triton X-100, 5% acetone, and up to 0.2% dimethylformamide) applied to the filter paper disk on top of 7-dram plastic vials. Two hours after application of test compound, 0.4 mL of citrated bovine blood was added to the filter paper disk. Each treatment level consisted of 10 replicate vials, each vial containing four adult flies. The experimental design is such that all groups are completely randomized within the whole, and the test vials are read in numerical order without knowledge of treatment history. Ivermectin is run as a positive control on every assay. The estimated LD_{90} is calculated from the motility of flies 24 h after exposure to test substances. The LD_{90} calculation is based on double arcsine transformed data fitted to straight lines with log-dose response curves. Data are reported in Table 1.

Sheep Assay. Anthelmintic efficacy of the compounds was determined in sheep raised helminth free and experimentally infected with the parasites listed in Table 2. When the infections were patent, the sheep were randomly assigned to a treatment or control group. Each compound was tested in one sheep while two sheep served as controls. The compounds were administered in a single oral dosage of 0.1 mg/kg. Control animals were given only the vehicle. Seven days after dosing, the animals were necropsied and the residual worm burdens were determined. The efficacies of the test compounds were recorded as described in footnote a of Table 2.

Estimated Mouse LD₅₀ Assay. Acute toxicity of the compounds was evaluated by calculating the LD_{50} for male

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Table 1.Avermectin Analogue Activities against BrineShrimp (A. salina) and S. calcitrans Adults

1-			brine shrimp	Stomoxys
compd ^a	X	R	IC ₁₀₀ ° (ng/mL)	LD_{90} (ppm)
1a	OH		356 ^c	40.9
1aD	OH		430°	29.5
1b	epi-AcNH		540°	5.3
2a	OH		760 ^c	33.0
2b	epi-AcNH		870	18.8
4c	epi-AcNH	Bz	870	2.1
4d	epi-AcNH	\mathbf{THP}	1500°	2.7
4e	epi-AcNH	Me	650	1.3
4f	epi-AcNH	CH ₂ OBn	870	1.7
4g	epi-AcNH	MEM	650	1.4
4gD	epi-AcNH	MEM	870	0. 9
4h	OH	MEM	325	3.6
4hD	OH	MEM	430	2.1
1i	OMEM		1300	75.6
2 i	OMEM		1730	8.7
4j	OMEM	MEM	1730	7.1
1 k	epi-AcNMe		650	13.1
2k	epi-AcNMe		650	9 .8
41	<i>epi-</i> AcNMe	MEM	870	1.8
1m	OMe		2600	31.7
4mD	OMe	Me	870	22.4

^a All derivatives are of avermectin unless noted as 22,23-dihydro by the addition of a **D** to the compound number. ^b Brine shrimp data as previously described (Blizzard et al., 1989), average of two assays unless otherwise noted. ^c Multiple assays.

compd	$H.c.^b$	Os.c.	<i>T.a</i> .	T.c.	C.c.	C.onc.	Oe.c.
1a	3	3	3	3	3	3	3
1aD	3	3	3	3	3	3	3
1b	3	3	3	3	3	3	3
2a	3	2	3	3	0	3	3
2b	0	0	0	0	0	0	0
4 c	1	0	0	0	0	0	0
4g	3	3	3	3	3	3	3
1i	3	0	1	0	0	0	3
2i	3	2	3	2	3	1	3
4j	3	0	3	3	3		3

^a Sheep data obtained as described under Experimental Procedures, all compounds tested at 0.1 mg/kg, efficacy as percent reduction from control: 0 = <50%, 1 = 51-75%, 2 = 76-95%, 3 = >95%. ^b H.c., Haemonchus contortus; Os.c., Ostertagia circumcincta; T.a., Trichostronglyus axei; T.c., Trichostrongylus colubriformis; C.c., Cooperia curticei; C.onc., Cooperia oncophora; Oe.c., Oesophagostomum columbianum.

CD-1 mice. Each compound was evaluated at several doses in treatment groups of five mice per dose allocated at random from a pool of mice. The compounds were administered orally in a vehicle via a calibrated syringe with a blunt-tipped needle. Seven days after dosing, the number of deaths was determined and the estimated LD₅₀ values were determined by the method of dose-pair responses. This method effectively emulates linear regression while minimizing the number of animals required.

General Procedure A: SeO₂ Oxidation. A 250-mL round-bottom flask fitted with a magnetic stirring bar, septum, and nitrogen inlet was charged with the desired avermectin analogue in CH₂Cl₂ (0.15 M). To the resulting clear solution was added 0.50 equiv of SeO₂ followed by 4 equiv of 90% *t*-BuOOH at room temperature. The resulting solution was stirred at room temperature for ~24 h. The reaction mixture was then concentrated by rotary evaporation and chromatographed with a CH₂Cl₂/MeOH solvent mixture on SiO₂ to provide the 4a-hydroxyavermectin analogue as a foam.

Data for 4a-Hydroxyavermectin B_1 (2a) (Mrozik, 1984a,b): 2.04 g (50%); $R_f = 0.17$ (94:6 CH₂Cl₂/MeOH). MS and combustion data are in the supplementary material.

Data for 4"-epi-(Acetylamino)-4"-deoxy-4a-hydroxyavermectin **B**₁ (2b): 6.10 g (50%); $R_f = 0.27$ (93:7 CH₂Cl₂/MeOH); ¹H NMR δ 5.86 (m, H₉), 5.78–5.63 (om, H₃, H₁₀, H₁₁, H₂₃, NH), 5.55 (dd, J = 9.9, 2.5 Hz, H₂₂), 5.38–5.25 (om, H₁₉, H_{1"}), 4.95 (m, H₁₅), 4.74 (d, J = 3.2 Hz, H₁'), 4.65 (m, H_{8a}), 4.55 (br d, J $= 5.0 \text{ Hz}, \text{H}_5), 4.38, (\text{dd}, J = 10.0, 3.2 \text{ Hz}, \text{H}_{4''}), 4.22 (\text{br s}, \text{H}_{4a}),$ 4.03 (m, $H_{5''}$), 3.97 (d, J = 6.3 Hz, H_6), 3.90 (br s, H_{13}), 3.90-3.72 (om, H_{17} , $H_{5'}$), 3.71–3.50 (om, $H_{3'}$, $H_{3''}$), 3.45 (d, J = 10Hz, H₂₅), 3.42 (s, OCH₃), 3.35 (s, OCH₃), 3.32 (m, H₂), 3.18 (t, J = 9.0 Hz, H₄'), 2.70 (br s, 2 × OH), 2.50 (m, H₁₂), 2.32–2.15 $(\text{om}, 2 \times H_{16}, H_{24}, H_{2'\text{eq}}), 2.07 \text{ (s, CH}_3\text{CO}), 2.05 - 1.95 \text{ (om}, H_{20\text{eq}}, H_{20\text{eq}}), 1.05 + 1.05 \text{ (om}, H_{20\text{eq}}), 1.05 \text{ (om}, H_{2$ $H_{2''eq}$, 1.75 (m, H_{18eq}), 1.65–1.40 (om, H_{20} , H_{26} , 2 × H_{27} , $H_{2'}$, $H_{2''}$), 1.48 (s, 3 × H_{14a}), 1.21 (d, J = 6.2 Hz, 3 × $H_{6'}$), 1.15 (d, J = 6.9 Hz, $3 \times H_{12a}$), 1.10 (d, J = 6.6 Hz, $3 \times H_{6''}$), 0.95-0.85(om, $3 \times H_{24a}$, $3 \times H_{26a}$, $3 \times H_{28}$, H_{18ax}); ¹³C NMR δ 173.2 (C₁), $170.9 [(C=O)NH], 140.2, 139.3 (C_{14}, C_8), 138.1 (C_{11}), 136.3 (C_{23}),$ $135.1\ (C_4),\ 127.7\ (C_{22}),\ 124.7\ (C_{10}),\ 120.7,\ 119.9,\ 118.3\ (C_3,\ C_9,\ 120.7,\ 119.9,\ 118.3\ (C_{10},\ C_{10}),\ 110.7,\ 119.9,\ 118.3\ (C_{10},\ C_{10}),\ 110.7,\ 119.9,\ 110.7,\ 119.9,\ 110.7,\ 119.9,\ 110.7,\ 119.9,\ 110.7,\ 119.9,\ 110.7,\ 119.9,\ 110.7,\ 119.9,\ 110.7,\ 119.9,\ 110.7,\ 119.9,\ 110.7,\ 119.9,\ 110.7,\ 119.9,\ 110.7,\ 119.9,\ 110.7,\ 119.9,\ 110.7,\ 119.9,\ 110.7,\ 119.9,\ 119.9,\ 119.9,\ 119.9,\ 119.9,\ 119.9,\ 119.9,\ 119.9,\ 119.9,\ 119.9,\ 119.9,\ 119$ $C_{15}),\,98.6\;(C_{1''}),\,95.8\;(C_{21}),\,94.9\;(C_{1'}),\,81.9\;(C_{13}),\,81.0\;(C_{4'}),\,80.6\;(C_{15}),\,81.0\;(C_{$ (C_7) , 79.3 $(C_{3'})$, 79.2 (C_6) , 74.9 (C_{25}) , 73.3 $(C_{3''})$, 68.6, 68.3 (C_{17}) , $C_{19}),\,68.4\,(C_{8a}),\,67.1,\,65.5\,(C_5,\,C_{5'},\,C_{5''}),\,64.4\,(C_{4a}),\,56.6,\,56.1\,(2$ \times OCH₃), 48.5 (C_{4"}), 45.6 (C₂), 40.5 (C₂₀), 39.8 (C₁₂), 36.5 (C₁₈), $35.2(C_{26}), 34.5(C_{2'}), 34.2(C_{16}), 31.8(C_{2''}), 30.6(C_{24}), 27.5(C_{27}),$ 23.4 (CH₃C=O), 20.2 (C_{12a}), 18.3 (C_{6'}), 17.0 (C_{6"}), 16.4 (C_{24a}), 15.1 (C_{14a}), 13.0 (C_{26a}), 12.0 (C₂₈); MS (FAB) 952 (M + Na, 4), 330 (26), 305 (22), 300 (20), 221 (20), 186 (100), 154 (82), 112 (34). Anal. Calcd for C₅₀H₇₅NO₁₅: C, 64.57; H, 7.95; N, 1.51. Found: C, 64.53; H, 7.95; N, 1.52.

Data for 4"-epi-N-Acetyl-4"-deoxy-N-(methylamino)-4a-hydroxyavermectin B₁ (2k): 1.35 g (46%); $R_f = 0.19$ (94:6 CH₂Cl₂/MeOH); ¹H NMR (selected data) δ 5.86 (m, H₉), 5.78– 5.63 (om, H₃, H₁₀, H₁₁, H₂₃), 5.55 (dd, J = 9.9, 2.5 Hz, H₂₂), 5.45 (d, J = 3.0 Hz, H_{1"}), 5.40–5.30 (m, H₁₉), 4.95 (m, H₁₅), 4.74 (d, J = 3.2 Hz, H_{1"}), 4.65 (m, H_{8a}), 4.55 (t, J = 5.0 Hz, H₅), 4.22 (br s, H_{4a}), 4.15 (m, H_{5"}), 3.95 (d, J = 6.3 Hz, H₆), 3.90 (br s, H₁₃), 3.42 (s, OCH₃), 3.35 (s, OCH₃), 3.32 (m, H₂), 3.18 (s, NCH₃), 2.85 (d, J = 5 Hz, OH); MS (FAB) 966 (M + Na, 80), 344 (90), 312 (100), 221 (95). Anal. Calcd for C₅₁H₇₇NO₁₆: C, 64.88; H, 8.22; N, 1.48. Found: C, 64.62; H, 8.33; N, 1.41.

RESULTS AND DISCUSSION

Chemistry. We initially focused on one $C_{4''}$ substituent (*epi*-acetylamino) (Mrozik et al., 1989; Mrozik, 1984a,b; Linn and Mrozik, 1991) and varied the groups at C_{4a} . Once the structure activity relationship was established, we examined the most desirable C_{4a} substituents with other $C_{4''}$ groups. In general, these compounds were prepared by first oxidizing at C_{4a} with SeO₂ as previously reported (Fisher and Mrozik, 1984; Mrozik, 1984a,b; Fraser-Reid et al., 1988), to provide alcohols **2a**, **2b**, **2i**, and **2k**. A series of protection and deprotection reactions selectively revealed C_{4a} OH in **3** for derivatization. Finally, complete deprotection provided the desired compounds (4) for testing (Scheme 1).

The oxidation reproducibly provided 4a-hydroxyavermectins in a yield of $\sim 50\%$; however, only very low yields were obtained with C5OH protected with either silyl or ester groups. Esters at $C_{4a}OH$ were directly introduced through a Mitsunobu reaction (Mrozik, 1984a,b; Mitsunobu, 1981; Hughes et al., 1988) without the aforementioned protection/deprotection sequence to provide compounds like 4c. Ethers, however, were prepared by silvlating with *t*-BuSiMe₂Cl (TBSCl) at C₅-OH and C_{4a} OH, desilylating C_{4a} OTBS, and trimethylsilylating C7OH and C4aOH with bis(trimethylsilyl)trifluoroacetamide (BSTFA) followed by hydrolysis of the primary silyl ether with aqueous HOAc to provide 3 from 2. The final silulation with BSTFA was necessary since the tertiary alcohol at C_7 was competitively alkylated by Me₃OBF₄, (2-methoxyethoxy)methyl chloride (MEMCl), and benzyloxymethyl chloride (BOMCl). Alkylation at C4aOH followed by deprotection with pyridine/HF (Trost et al., 1983) provided the desired compounds (4) for testing. The methyl ethers were prepared by alkylation with Me₃OBF₄. This reagent competitively alkylates the $C_{4''}$ acetylamino group, so





A: a) TBSCI, imidazole, DMF, rt; b) HF/pyr, THF, rt; c) BSTFA, DMF, rt; d) HOAc, H₂O

the reaction was run to alkylate both the amide and alcohol functional groups. Upon complete alkylation, the reaction mixture was quenched with water to provide a $C_{4''}$ amino group that was reacylated with Ac_2O .

Silvlation at $C_{4''}OH$ (and C_7OH) was sufficiently slower than silvlation at C_5OH and $C_{4a}OH$ (see structure 2), so that good yields of either silvlated (allowing the reaction to run for extended periods of time) or free alcohol (stopping the reaction when C_5OH was silvlated) at $C_{4''}$ could be obtained. Identical groups could therefore be introduced at $C_{4a}OH$ and at $C_{4''}OH$ by selectively silylating $C_{4a}OH$ and C_5OH and then selectively hydrolyzing the primary silyl group to provide a C₅ monosilyl ether-C4a, C4", C7 triol. Pertrimethylsilylation followed by aqueous hydrolysis gave a C_{4a} , $C_{4''}$ diol ready for alkylation. Deprotection once again provided the desired compounds (4) for testing. The analogues with a hydroxyl at $C_{4''}$ were prepared by silvlating $C_{4a}OH$, $C_{4''}$ -OH, and C_5OH and then selectively hydrolyzing the primary silyl ether at C_{4a} . As described above, the alcohols at C_7 and C_{4a} were then trimethylsilylated and the primary silyl ether was hydrolyzed with aqueous HOAc. Alkylation followed by hydrolysis with pyridine/ HF vielded the desired compounds (4).

A tetrahydropyran (THP) substituent was attached on $C_{4a}OH$ through a slightly different sequence as shown in Scheme 2. Silylation of $C_{4a}OH$ followed by esterification of C_5OH provided 5. Desilylation at C_{4a} provided the primary alcohol, 6, for derivatization with 3,4-dihydro-2*H*-pyran (DHP). The ester at C_5 was then removed by methanolysis to provide 4d for testing.

Biological Evaluation. The new avermectin ana-

logues were evaluated for activity in two in vitro assays against brine shrimp (Blizzard et al., 1989) (Artemia salina) and flies (Stomoxys calcitrans adults, Table 1). As shown in Table 1, derivatization at C_{4a} usually increased activity against flies relative to the parent $C_{4''}$ substituted avermectins. Nonbranched ethers (4e-g)were more active than free hydroxyls (2a, 2b, 2i, 2k), an ester (4c), or branched ether (4d). Saturating the $C_{22}-C_{23}$ double bond of the avermettin analogues produced compounds with increased activity against flies (compare 1aD, 4gD, and 4hD to 1a, 4g, and 4h). The more easily installed MEM group appeared to be as good as or better than a simple methyl substituent for increasing potency against flies (compare 4g, 4mD to 4e, 4j). Both of the acylamino groups at $C_{4''}$ and the groups at C_{4a} served to increase potency against flies relative to 1a. Combined on the same molecule, they provided even more potent compounds (4g and 4l), with the C_{4a} substituent providing a larger impact on efficacy. No strong correlation existed between the brine shrimp and fly data.

Selected compounds were further evaluated for *in vivo* activity against several parasites in sheep (Table 2) (Chabala et al., 1980). As shown in Table 2, a free hydroxyl or ester group at C_{4a} generally reduced antiparasitic activity in sheep (compare 2a, 2b, and 4c to 1a, 1b, and 1b). However, substitution with the chemically less reactive (and less polar) MEM group restored the activity to levels at or above the parent avermectin's levels (compare 4g and 4j to 1b and 1i). Finally, analysis of these avermectin analogues in a mouse LD_{50} assay showed that C_{4a} substitution generally provided compounds that were potentially safer in mammals than

Scheme 2



Table 3.	Acute Toxicity	of Avermectin	Analogues in
Mice ^a			

compd	$LD_{50} \ (mg/kg)$	compd	$LD_{50}(mg\!/\!kg)$
 1a	19	4c	≫560
1aD	27	4e	175
1b	24	4g	> 560
2a	205	4gD	≫320
		4ĥ	≫100
		4hD	310

 $^{\alpha}$ Estimated mouse LD_{50} as described under Experimental Procedures.

the parent derivatives, with the MEM ether (4g) superior to a methyl ether (4e) (Table 3).

Conclusion. Oxidation of avermectin analogues at C_{4a} followed by derivatization of the added oxygen provides safer compounds, as judged by mouse LD_{50} data, with substantial improvement in activity against flies. The antiparasitic activity of avermectin can be fully retained (see **4g**) by forming a MEM ether with the added C_{4a} oxygen while increasing activity against flies. The improved safety profile and efficacy against flies may potentially expand the use of avermectins.

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Supplementary Material Available: A procedure for 4c, general procedures B-F, selected mass spectral, combustion analytical, and transcribed and assigned NMR data for 2a, 4c, 4e, 4f, 4g, 4gD, 4h, 4hD, 2i, 4j, 4l, and 4mD; procedures and data for 4a-O-TBS-2b, 5, 6, and 4d (14 pages). Ordering information is given on any current masthead page.

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